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How to Build an Atom - Phet  
Simulation GAME OF BUILD AN  
ATOM The \"Build an Atom\" Phet Lab,

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Instruction Day 9 - Building an Atom  
PhET walkthrough Mrs. KJ Explains:  
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Structure of an Atom

Unit 1 Lesson 6 Building Atoms Lewis Diagrams Made Easy: How to Draw Lewis Dot Structures  
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Answer Key.pdf

Responses will vary, but here is what you should see&mldr;. Adding protons

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changes the identity of the atom  
adding one moves from right to left on  
the Periodic Table. You can get the  
atom to be stable/unstable by  
adjusting the number of neutrons  
Adding electrons will change it from a  
+ion to neutral atom, to ion depending  
on how many electrons you add



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Adding electrons you can only add 2 to the first orbital and 8 to the second b) What particle(s) are found in the center of the atom?

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BUILD AN ATOM ATOM SCREEN 1.

Explore the Build an Atom simulation.

a) List two things you observed in the simulation. Proton, neutron and

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electron are different colors The amount of these three changes the element b) What particle(s) are found in the center of the atom? Protons and neutrons 1. Play until you discover which particle(s) determine(s) the name of the element you build.

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identify an element, its

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hydrogen2 and hydrogen3 are both isotopes of hydrogen1 magnesium25 is an isotope of magnesium24 carbon-13 and carbon14 are both isotopes of

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carbon12 (carbon12 1 is an anion of carbon12, carbon12 +1 is a cation of carbon12) anions are negative ions with more electrons than protons cations are positive ions with fewer electrons than protons (the Gallium examples are also ions) 5 BUILD AN ATOM 6 BUILD AN ATOM

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the mass and charge. Predict how addition or subtraction of a ...

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“Build An Atom” Simulation – Build an Ion Learning Goals: 1. Draw models that show atomic structure. 2. Use information about the number of protons, neutrons, and electrons to

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identify an element, its position on the periodic table. 3. Predict how changing the number protons and electrons will change the element and its charge.

"Build An Atom" Simulation – Build an Ion

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you should see&mldr;. Adding protons changes the identity of the atom adding one moves from right to left on the Periodic Table. You can get the atom to be stable/unstable by adjusting the number of neutrons Adding electrons will change it from a +ion to neutral atom, to ion depending

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on how many electrons you add  
Adding electrons you can only add 2 to the first orbital and 8 to the second b)  
What particle(s) are found in the center of the atom?

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Explore the Build an Atom simulation.  
a) List two things you observed in the

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simulation. Proton, neutron and electron are different colors The amount of these three changes the element b) What particle(s) are found in the center of the atom? Protons and neutrons 1. Play until you discover which particle(s) determine(s) the name of the element you build.

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protons, neutrons, and electrons to identify an element, its

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hydrogen2 and hydrogen3 are both isotopes of hydrogen1  
magnesium25 is an isotope of magnesium24  
carbon-

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13 and carbon14 are both isotopes of carbon12 (carbon12 1 is an anion of carbon12, carbon12 +1 is a cation of carbon12) anions are negative ions with more electrons than protons cations are positive ions with fewer electrons than protons (the Gallium examples are also ions) 5 BUILD AN

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Then play a game to test your ideas!

Sample Learning Goals Use the number of protons, neutrons, and electrons to draw a model of the atom,

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